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Structure of 9-Bromo-3-endo-methylcamphor*

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C(11) Br' O' C(1') C(2')

C(6') C(7') C(8') C(9') C(10') C(11')

Abstract. $C_{11}H_{17}BrO$, $M_r = 245 \cdot 16$, orthorhombic, $P2_12_12$, $a = 14 \cdot 887$ (1), $b = 21 \cdot 590$ (2), $c = 7 \cdot 0528$ (3) Å, $V = 2266 \cdot 8$ (3) Å³, Z = 8 (2 molecules per asymmetric unit), $D_x = 1 \cdot 436$ (1) g cm⁻³, Mo Ka, $\lambda \alpha_1 = 0 \cdot 70930$ Å, $\mu = 36$ cm⁻¹, F(000) = 1008, T = 295 K, R = 0.058 for 1178 reflections. The geometry and dimensions of the molecule are similar to those in related materials; the C(1)-C(7)-C(4) bridgehead angle is $92 \cdot 7$ (7)°, and the mean C-Br bond distance is $1 \cdot 98$ (1) Å.

Experimental. Colourless crystals, dimensions $0.15 \times$ 0.12×0.60 mm, faces {010}, {001}, {120}; Enraf-Nonius CAD-4F diffractometer; lattice parameters a from 25 reflections with $\theta = 15-21^{\circ}$. Intensities for $\theta \le 25^{\circ}$, *hkl*: 0 to 17, 0 to 25, 0 to 8; ω -2 θ scan, ω scan width $(0.65 + 0.35 \tan \theta)^{\circ}$ at $1.3-10^{\circ} \min^{-1}$, Br extended 25% on each side for background measure-0 ment; three standard reflections (no decay); Lp and C(1) C(2) intensity absorption corrections, transmission factors C(3) 0.38-0.57; 2304 independent reflections measured, C(4) 1178 with $I \ge 3\sigma(I), \sigma^2(I) = S + 4(B_1 + B_2) + (0.04S)^2$, C(5) C(6) $S = \text{scan}, B_1, B_2 = \text{background counts. Structure by}$ C(7) Patterson and Fourier methods; refined by full-matrix C(8) C(9) least squares on F; H atoms in calculated positions; C(10)





grams; final R = 0.058, wR = 0.076 for 1178 reflections, S = 3.12, 235 parameters; R = 0.121 for all 2304 reflections; $\Delta/\sigma = 0.06$ (mean), 0.29 (max.); max. final difference density -0.8 to +0.6 eÅ⁻³. Final positional parameters are in Table 1, bond distances Table 1. Final positional (fractional. $\times 10^4$. Br $\times 10^5$)

enantiomorph gave significantly poorer agreement

(R = 0.079 vs 0.071 before final refinement); w =

 $1/\sigma^2(F)$; scattering factors from International Tables

for X-ray Crystallography (1974); locally written, or

locally modified versions of standard computer pro-

able	1. Final pos	itional (fra	ctional, × 1	$0^{\circ}, Br \times 10^{\circ})$	
nd	equivalent	isotropic	thermal	parameters	
	$(U \times 10^3 \text{ Å}^2)$ with e.s.d.'s in parentheses				

x	У	z	U_{eq}^*
36185 (15)	25692 (9)	42254 (24)	89
5701 (8)	1496 (5)	11381 (19)	157
5286 (11)	1524 (6)	8040 (19)	45
5460 (10)	1801 (8)	9919 (22)	53
5399 (9)	2478 (7)	9888 (23)	48
5155 (9)	2564 (6)	7773 (17)	40
5964 (10)	2443 (9)	6583 (22)	66
6136 (10)	1754 (8)	6840 (24)	64
4536 (9)	1993 (7)	7392 (20)	46
3665 (11)	1942 (7)	8540 (21)	62
4305 (11)	1876 (7)	5278 (21)	60
5091 (14)	826 (8)	7872 (29)	93
6199 (13)	2819 (9)	10703 (24)	93
36392 (16)	39594 (9)	106580 (24)	89
2172 (8)	5521 (6)	3693 (17)	86
2222 (9)	5189 (7)	6994 (21)	46
2605 (11)	5331 (7)	5030 (24)	54
3586 (11)	5210 (6)	5112 (17)	46
3716 (9)	5033 (6)	7193 (19)	42
3594 (14)	5597 (7)	8407 (22)	64
2618 (13)	5749 (8)	8145 (24)	65
2846 (9)	4650 (6)	7569 (17)	40
2745 (12)	4085 (7)	6350 (23)	64
2683 (10)	4507 (7)	9604 (20)	59
1258 (14)	5126 (9)	7013 (28)	89
4165 (13)	5760 (10)	4361 (29)	95

* $U_{eq} = \frac{1}{3}$ trace of diagonalized U tensor.

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and angles in Table 2, and other data have been deposited.* The two independent molecules (Fig. 1) have similar structure and conformation, the molecular geometry being generally close to that observed for other camphor derivatives (Rettig & Trotter, 1986). Absolute configuration given by Hutchinson & Money (1984).

* Lists of anisotropic thermal parameters, H-atom positions, torsion angles and structure factors have been deposited with the British Library Lending Division as Supplementary Publication No. SUP 43044 (16 pp.). Copies may be obtained through The Executive Secretary, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

Table 2. Bond lengths (Å) and angles (°), with e.s.d.'sin parentheses

BrC(9)	1.957 (14)	Br'-C(9')	1.994 (15)
O-C(2)	1.27 (2)	O'-C(2')	1.21 (2)
C(1) - C(2)	1.48 (2)	C(1') - C(2')	1.53 (2)
C(1)-C(6)	1.60 (2)	C(1') - C(6')	1.57 (2)
C(1)-C(7)	1.58 (2)	C(1')-C(7')	1.54 (2)
C(1)C(10)	1.54 (2)	C(1')-C(10')	1.44 (2)
C(2)-C(3)	1.46 (2)	C(2') - C(3')	1.48 (2)
C(3)–C(4)	1.55 (2)	C(3')-C(4')	1.53 (2)
C(3)-C(11)	1-51 (2)	C(3') - C(11')	1.56 (2)
C(4)C(5)	1.49 (2)	C(4')-C(5')	1.50 (2)
C(4) - C(7)	1.56 (2)	C(4') - C(7')	1.56 (2)
C(5)-C(6)	1.52 (2)	C(5') - C(6')	1.50 (3)
C(7) - C(8)	1.53 (2)	C(7')-C(8')	1.50 (2)
C(7)-C(9)	1.55 (2)	C(7') - C(9')	1.49 (2)
C(2)C(1)C(6)	102.1 (12)	C(2')-C(1')-C(6')	100.0 (12)
C(2)-C(1)-C(7)	97-1 (12)	C(2')-C(1')-C(7')	99.5 (11)
C(2)-C(1)-C(10)	120-0 (13)	C(2')-C(1')-C(10')	113.5 (15)
C(6)-C(1)-C(7)	101.9 (11)	C(6')–C(1')–C(7')	102.7 (11)
C(6) - C(1) - C(10)	114-4 (14)	C(6')-C(1')-C(10')	116-2 (14)
C(7)-C(1)-C(10)	118-3 (13)	C(7')-C(1')-C(10')	121.6 (14)
O-C(2)-C(1)	124-4 (15)	O'-C(2')-C(1')	125-0 (14)
O - C(2) - C(3)	123 (2)	O'-C(2')-C(3')	127.8 (14)
C(1)-C(2)-C(3)	112-4 (12)	C(1')-C(2')-C(3')	107-2 (13)
C(2)-C(3)-C(4)	98-6 (12)	C(2')-C(3')-C(4')	101.9 (12)
C(2)-C(3)-C(11)	115-5 (13)	C(2')-C(3')-C(11')	113-4 (13)
C(4) - C(3) - C(11)	119-5 (13)	C(4')–C(3')–C(11')	116-5 (14)
C(3) - C(4) - C(5)	109-4 (12)	C(3')-C(4')-C(5')	109.3 (11)
C(3) - C(4) - C(7)	102-1 (11)	C(3')-C(4')-C(7')	101-0 (11)
C(5)-C(4)-C(7)	103.9 (11)	C(5')-C(4')-C(7')	103-5 (11)
C(4) - C(5) - C(6)	103-9 (13)	C(4')–C(5')–C(6')	103.0 (13)
C(1) - C(6) - C(5)	103 5 (12)	C(1')-C(6')-C(5')	104.9 (13)
C(1)-C(7)-C(4)	92.3 (10)	C(1')–C(7')–C(4')	93-1 (10)
C(1)-C(7)-C(8)	113.5 (12)	C(1')–C(7')–C(8')	113.8 (12)
C(1)-C(7)-C(9)	109-4 (11)	C(1')–C(7')–C(9')	108-2 (12)
C(4)–C(7)–C(8)	117.7 (11)	C(4')–C(7')–C(8')	114-6 (11)
C(4) - C(7) - C(9)	115.1 (13)	C(4')-C(7')-C(9')	114.1 (12)
C(8) - C(7) - C(9)	108.0 (12)	C(8')-C(7')-C(9')	111.6 (13)
Br-C(9)-C(7)	110.9 (10)	Br'-C(9')-C(7')	111.5 (10)



Fig. 1. Stereoviews of the two independent 9-bromo-3-endomethylcamphor molecules.

Related literature. The crystal structure analysis was undertaken to assist in the characterization of the products of bromination of camphor derivatives (Hutchinson & Money, 1984).

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